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review. liquid that
dissolves a gas or
solid or the solution
component in greater
amount. when a
system in equilibrium
is altered by a change
of temperature,
pressure, or
concentration

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variable, the system shifts in equilibrium composition in a way that tends to counteract this change of variable.

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Solutions Name: Carly

1) Does a solution have to involve a liquid? Explain your answer. No; solutions may exist as gases, liquids, or solids. 2) Explain why a suspension is considered a

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heterogeneous

mixture. The particles are so large that they settle out unless the mixture is constantly stirred or agitated.

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Solutions Name ...

Modern Chemistry 5

Solutions CHAPTER

12 REVIEW Solutions

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SECTION 2 SHORT

ANSWER Answer the following questions

in the space

provided. 1. The

following are

statements about the dissolving process.

Explain each one at the molecular level. a.

Increasing the

pressure of a solute

gas above a liquid

solution increases the

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solubility of the gas
in the liquid.

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SECTION 1 SHORT

ANSWER Answer the following questions in the space

provided. 1. Match the type of mixture on the left to its representative particle diameter on the right. c solutions (a) larger than 1000

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1 nm a suspensions (b)
1 nm to 1000 nm b
colloids (c)

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CHAPTER 12 REVIEW
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SHORT ANSWER

Answer the following
questions in the
space provided. 1.

Match the type of

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Review

mixture on the left to its representative particle diameter on the right. c solutions (a) larger than 1000 nm a suspensions (b) 1 nm to 1000 nm b colloids (c) smaller than 1 nm 2. Identify the solvent in each of the following examples:

mc06se cFMsr i-vi

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Review_and_Discussi
on 13 : 5. How does ...

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Modern Chemistry
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Chapter 12-Solutions
Section 1- Types of
Mixtures Solutions
are homogeneous
mixtures of two or
more substances in a
single phase. Soluble
describes a substance
as capable of being
dissolved. Solvent is

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the dissolving
medium in a solution.

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makes a thorough

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a compound is not an

electrolyte, identify

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and calculate moles,
grams, and
molalilities in various

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